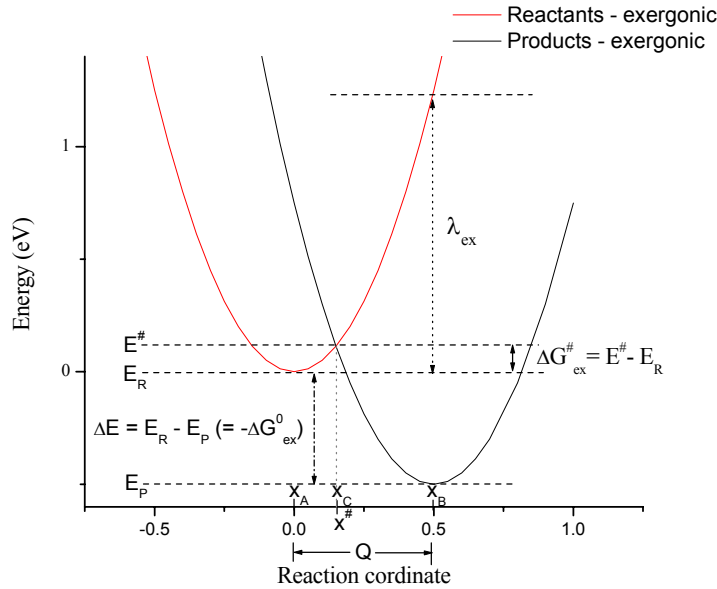


Marcus equation solutions for exergonic reaction



The parabolic equation for the reactants:

$$E = k_R(x - x_A)^2 + E_R \quad (1)$$

The parabolic equation for the products:

$$E = k_P(x - x_B)^2 + E_P \quad (2)$$

Solve for the energy of activation for the exergonic reaction, $\Delta G_{ex}^\#$, using equation (1), points $\{x_C, E^\#\}$, $\{x_B, \lambda_{ex} + E_R\}$ and equation (2), point $\{x_C, E^\#\}$:

i) Substitution of point $\{x_C, E^\#\}$ into equation (1):

$$E^\# = k_R(x_C - x_A)^2 + E_R$$

$$\text{Substitute : } x_C = x^\# + x_A$$

$$E^\# = k_R((x^\# + x_A) - x_A)^2 + E_R$$

$$E^\# = k_R(x^\#)^2 + E_R$$

$$\text{Substitute : } \Delta G_{ex}^\# = E^\# - E_R$$

$$E^\# - E_R = \Delta G_{ex}^\# = k_R(x^\#)^2$$

$$x^\# = \left(\frac{\Delta G_{ex}^\#}{k_R}\right)^{\frac{1}{2}}$$

ii) Substitution of point $\{x_B, \lambda_{ex} + E_R\}$ into equation (1):

$$\lambda_{ex} + E_R = k_R(x_B - x_A)^2 + E_R$$

$$\text{Substitute : } Q = x_B - x_A$$

$$\lambda_{ex} = k_R(Q)^2$$

$$Q = \left(\frac{\lambda_{ex}}{k_R}\right)^{\frac{1}{2}}$$

iii) Substitution of point $\{x_C, E^\#\}$ into equation (2):

$$E^\# = k_P (x_C - x_B)^2 + E_P$$

$$\text{Substitute : } x_C = x^\# - x_A$$

$$E^\# - E_P = k_P ((x^\# - x_A) - x_B)^2$$

$$\text{Substitute : } Q = x_B - x_A$$

$$E^\# - E_P = k_P (x^\# - Q)^2$$

$$\text{Substitute : } E_P = E_R - \Delta E$$

$$E^\# - (E_R - \Delta E) = k_P (x^\# - Q)^2$$

$$\text{Substitute : } \Delta G_{ex}^\# = E^\# - E_R$$

$$\Delta G_{ex}^\# + \Delta E = k_P (x^\# - Q)^2$$

From(i) & (ii) above

$$x^\# = \left(\frac{\Delta G_{ex}^\#}{k_R} \right)^{\frac{1}{2}}$$

$$Q = \left(\frac{\lambda_{ex}}{k_R} \right)^{\frac{1}{2}}$$

$$\Delta G_{ex}^\# + \Delta E = k_P \left(\left(\frac{\Delta G_{ex}^\#}{k_R} \right)^{\frac{1}{2}} - \left(\frac{\lambda_{ex}}{k_R} \right)^{\frac{1}{2}} \right)^2$$

$$\Delta G_{ex}^\# + \Delta E = k_P \left(\left(\frac{\Delta G_{ex}^\#}{k_R} \right) - 2 \left(\frac{\lambda_{ex}}{k_R} \frac{\Delta G_{ex}^\#}{k_R} \right)^{\frac{1}{2}} + \left(\frac{\lambda_{ex}}{k_R} \right) \right)$$

$$\text{Assume : } k_P = k_R$$

$$\Delta G_{ex}^\# + \Delta E = \left(\Delta G_{ex}^\# - 2(\lambda_{ex} \Delta G_{ex}^\#)^{\frac{1}{2}} + \lambda_{ex} \right)$$

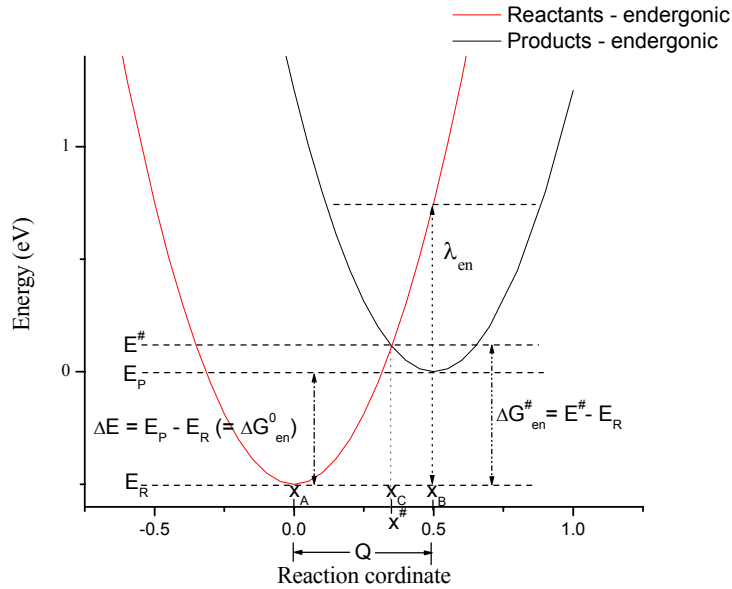
$$\Delta E = -2(\lambda_{ex} \Delta G_{ex}^\#)^{\frac{1}{2}} + \lambda_{ex}$$

$$\Delta G_{ex}^\# = \frac{(\lambda_{ex} - \Delta E)^2}{4\lambda_{ex}}$$

$$\text{Substitute : } \Delta E = -\Delta G_{ex}^0$$

$$\Delta G_{ex}^\# = \frac{(\lambda_{ex} + \Delta G_{ex}^0)^2}{4\lambda_{ex}}$$

Marcus equation solutions for endergonic reaction



The parabolic equation for the reactants:

$$E = k_R(x - x_A)^2 + E_R \quad (1)$$

The parabolic equation for the products:

$$E = k_P(x - x_B)^2 + E_P \quad (2)$$

Solve for the energy of activation for the endergonic reaction, $\Delta G_{en}^\#$, using equation (1), points $\{x_C, E^\#\}$, $\{x_B, \lambda_{en} + E_R\}$ and equation (2), point $\{x_C, E^\#\}$:

iv) Substitution of point $\{x_C, E^\#\}$ into equation (1):

$$E^\# = k_R(x_C - x_A)^2 + E_R$$

$$\text{Substitute : } x_C = x^\# + x_A$$

$$E^\# = k_R((x^\# + x_A) - x_A)^2 + E_R$$

$$E^\# = k_R(x^\#)^2 + E_R$$

$$\text{Substitute : } \Delta G_{en}^\# = E^\# - E_R$$

$$E^\# - E_R = \Delta G_{en}^\# = k_R(x^\#)^2$$

$$x^\# = \left(\frac{\Delta G_{en}^\#}{k_R}\right)^{\frac{1}{2}}$$

v) Substitution of point $\{x_B, \lambda_{en} + E_R\}$ into equation (1):

$$\lambda_{en} + E_R = k_R(x_B - x_A)^2 + E_R$$

$$\text{Substitute : } Q = x_B - x_A$$

$$\lambda_{en} = k_R(Q)^2$$

$$Q = \left(\frac{\lambda_{en}}{k_R}\right)^{\frac{1}{2}}$$

vi) Substitution of point $\{x_C, E^\#\}$ into equation (2):

$$E^\# = k_p(x_C - x_B)^2 + E_p$$

$$\text{Substitute : } x_C = x^\# + x_A$$

$$E^\# - E_p = k_p((x^\# + x_A) - x_B)^2$$

$$\text{Substitute : } Q = x_B - x_A$$

$$E^\# - E_p = k_p(x^\# - Q)^2$$

$$\text{Substitute : } E_p = E_R + \Delta E$$

$$E^\# - (E_R + \Delta E) = k_p(x^\# - Q)^2$$

$$\text{Substitute : } \Delta G_{en}^\# = E^\# - E_R$$

$$\Delta G_{en}^\# - \Delta E = k_p(x^\# - Q)^2$$

From(i) & (ii) above

$$x^\# = \left(\frac{\Delta G_{en}^\#}{k_R}\right)^{\frac{1}{2}}$$

$$Q = \left(\frac{\lambda_{en}}{k_R}\right)^{\frac{1}{2}}$$

$$\Delta G_{en}^\# - \Delta E = k_p \left(\left(\frac{\Delta G_{en}^\#}{k_R}\right)^{\frac{1}{2}} - \left(\frac{\lambda_{en}}{k_R}\right)^{\frac{1}{2}} \right)^2$$

$$\Delta G_{en}^\# - \Delta E = k_p \left(\left(\frac{\Delta G_{en}^\#}{k_R}\right) - 2\left(\frac{\lambda_{en}}{k_R} \frac{\Delta G_{en}^\#}{k_R}\right)^{\frac{1}{2}} + \left(\frac{\lambda_{en}}{k_R}\right) \right)$$

$$\text{Assume : } k_p = k_R$$

$$\Delta G_{en}^\# - \Delta E = (\Delta G_{en}^\# - 2(\lambda_{en} \Delta G_{en}^\#)^{\frac{1}{2}} + \lambda_{en})$$

$$-\Delta E = -2(\lambda_{en} \Delta G_{en}^\#)^{\frac{1}{2}} + \lambda_{en}$$

$$\Delta G_{en}^\# = \frac{(\lambda_{en} + \Delta E)^2}{4\lambda_{en}}$$

$$\text{Substitute : } \Delta E = \Delta G_{en}^0$$

$$\Delta G_{en}^\# = \frac{(\lambda_{en} + \Delta G_{en}^0)^2}{4\lambda_{en}}$$

$$\Delta G_{en}^\# = \frac{(\Delta G_{en}^0 + \lambda_{en})^2}{4\lambda_{en}}$$